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UHTC Composites: Processing, Performance and Future

Jon Binner
University of Birmingham

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Ultra-High Temperature Ceramics: Processing, Properties and Applications

Jon Binner

Deputy Head of the Engineering & Physical Sciences College
and Professor of Ceramic Science & Engineering
University of Birmingham, UK

Applications

- Potential future advanced aerospace vehicles
 - Hypersonic aviation
 - Re-usable atmospheric re-entry vehicles
 - Air-breathing hypersonic missile systems
- Nuclear
 - Advanced reactor designs
- Other niche applications
 - High temperature electrodes
 - Molten metal handling

**Require new materials with
>2000°C capability**

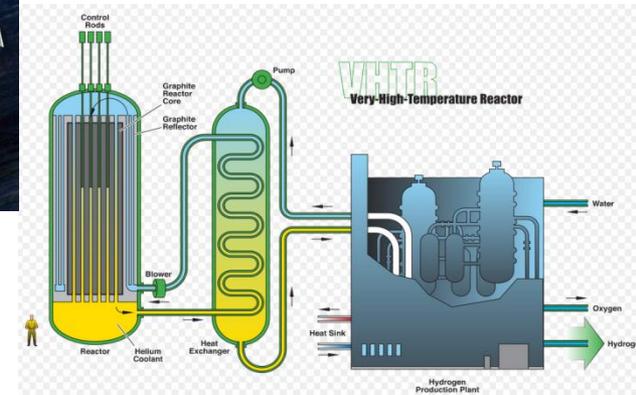


X-43B Hypersonic Concept
Image Courtesy of NASA



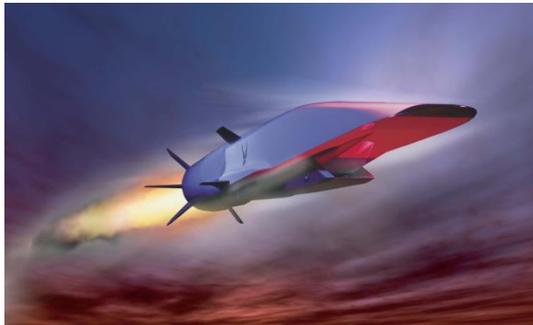
Hypersonic scramjet
concept missile

Very high temperature
nuclear reactor (VHTR)

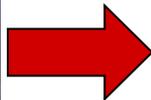


Key Issues for Implementation

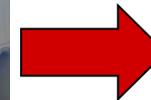
- **Performance:** Can improvements be made to resistance to oxidation, ablation, thermal shock, thermal cycling and creep?
- **Properties:** Is component behaviour representative of the inherent material properties?
- **Fabrication:** Can parts be made to near net shape at reasonable cost?



X-51 Concept
Image Courtesy of NASA



X-51 Test Vehicle
Image Courtesy of NASA



Can we move
from concept to
an operational
hypersonic flight
vehicle?

Materials Systems for Extreme Environments

XMat



Jon Binner, University of Birmingham

Bill Lee, Imperial College London

Mike Finnis, Imperial College London

Mike Reece, Queen Mary London



EPSRC

Engineering and Physical Sciences
Research Council



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Imperial College
London

 Queen Mary
University of London



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XMat

Materials Systems for Extreme
Environments

Background

The overall objective is to establish in the UK the capability to discover and understand new materials (not just UHTCs) that can operate under increasingly extreme conditions, thus enabling a wide range of new technologies.

The vision is to develop the required understanding of how the processing, microstructures and properties of materials systems operating in extreme environments interact to the point where materials with the required performance can be designed and then manufactured.

Funding is provided by EPSRC via the Programme Grant scheme and is valued at £4.2M (A\$8.1M) over 5 years. It formally started on the 1st February 2013.



Partners

AFRL	Morgan Ceramics
AWE	NASA
CERAM	NNL
Culham	NPL
DSTL	Powders
ESA	Reaction Engines
FCT	Sintec
ISTEC	Teledyne
Kerneos	Tokamak
Limoges	TWI
Missouri S&T	Vesuvius

Additional programmes to date valued at nearly £500k (A\$960k).

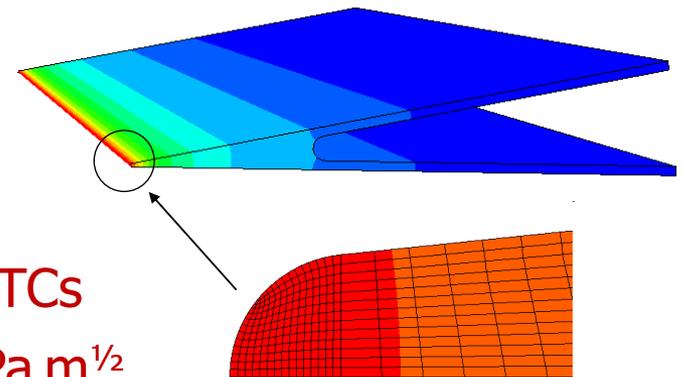
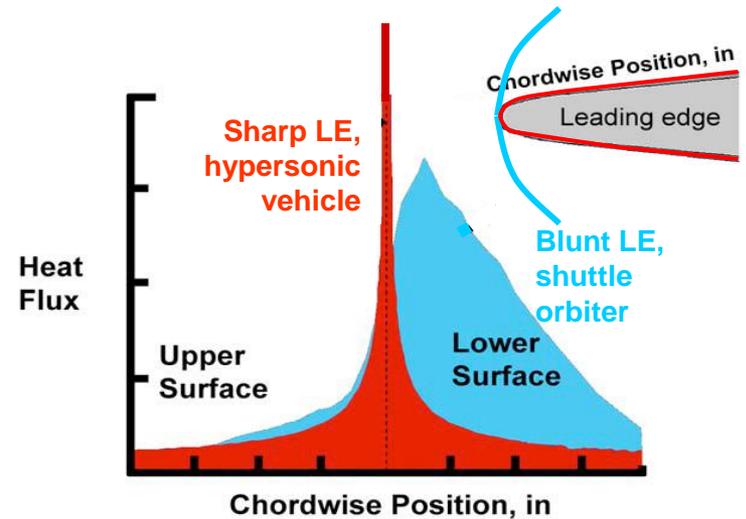
Materials for Sharp Leading Edges

Materials need to withstand:

- Very high heat fluxes over small areas
- Very high temperatures, oxidation & erosion
- Very high temperature gradients

ZrB₂/HfB₂-based ceramics offer:

- Ultra-high temperature capability
- High thermal conductivity
- Some oxidation resistance – best of the UHTCs
- Low fracture toughness – typically 3 – 4 MPa m^{1/2}



Steep temperature gradient with high temperature at tip

Materials for Sharp Leading Edges

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HfB₂ gel cast sample



Carbon-Carbon Based Composites for UHTC Applications

Advantages of CC composites

- Excellent high temperature strength
- Light weight
- Low coefficient of thermal expansion
- Good ablation resistance

Disadvantages

- Poor oxidation resistance above 500°C

The original goal of the project was to produce UHTC powder impregnated C_f-based composites that would withstand as high a temperature as possible for as long as possible.

Composite Manufacture

- ZrB₂
- HfC
- **HfB₂**
- ZrB₂+20vol%SiC
- ZrB₂+20vol%SiC+10vol%LaB₆

UHTC - phenolic slurry

Surface Transforms

2.5 D Cf preform

Vacuum impregnation

Drying

Curing

Pyrolysis

C-CVI



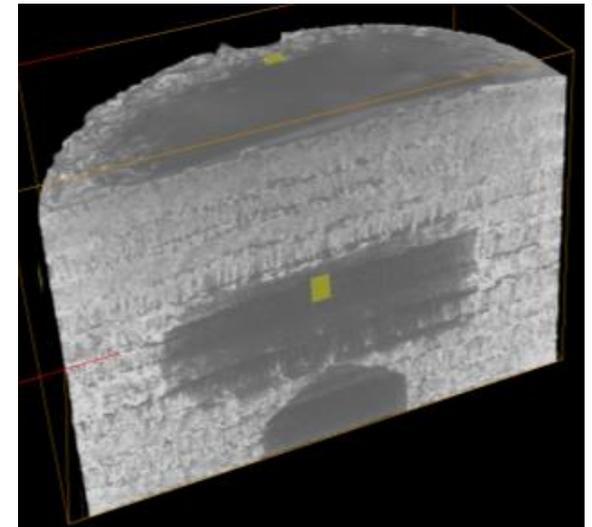
Preform 2.5D



Impregnation & curing



Pyrolysis



Impregnated carbon preform

Oxyacetylene Torch Testing (OAT)

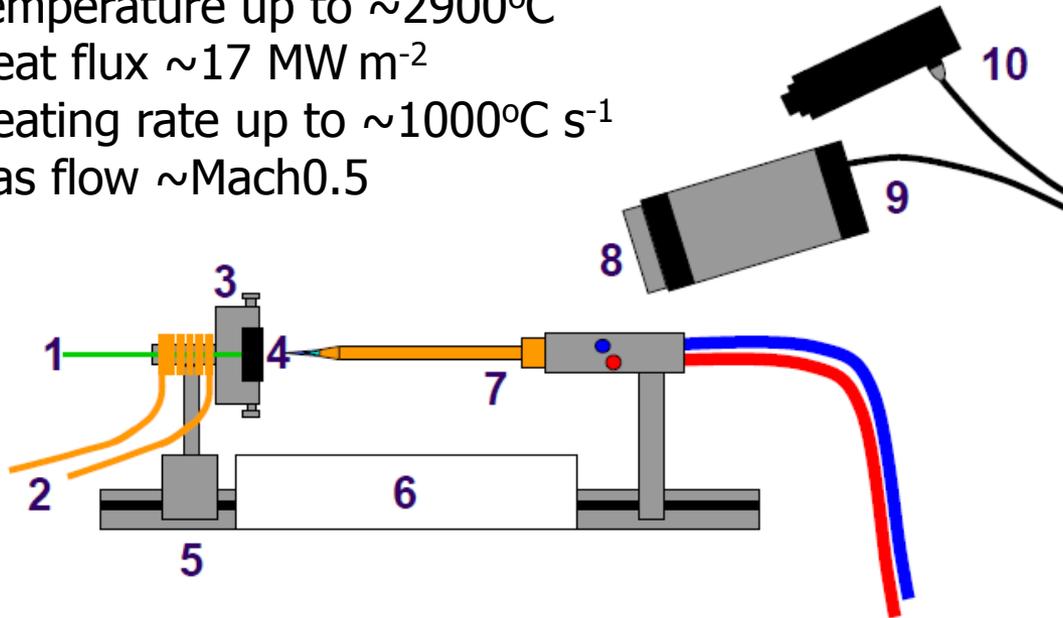
Sample typically 10 mm from flame tip

Temperature up to $\sim 2900^{\circ}\text{C}$

Heat flux $\sim 17 \text{ MW m}^{-2}$

Heating rate up to $\sim 1000^{\circ}\text{C s}^{-1}$

Gas flow $\sim \text{Mach}0.5$

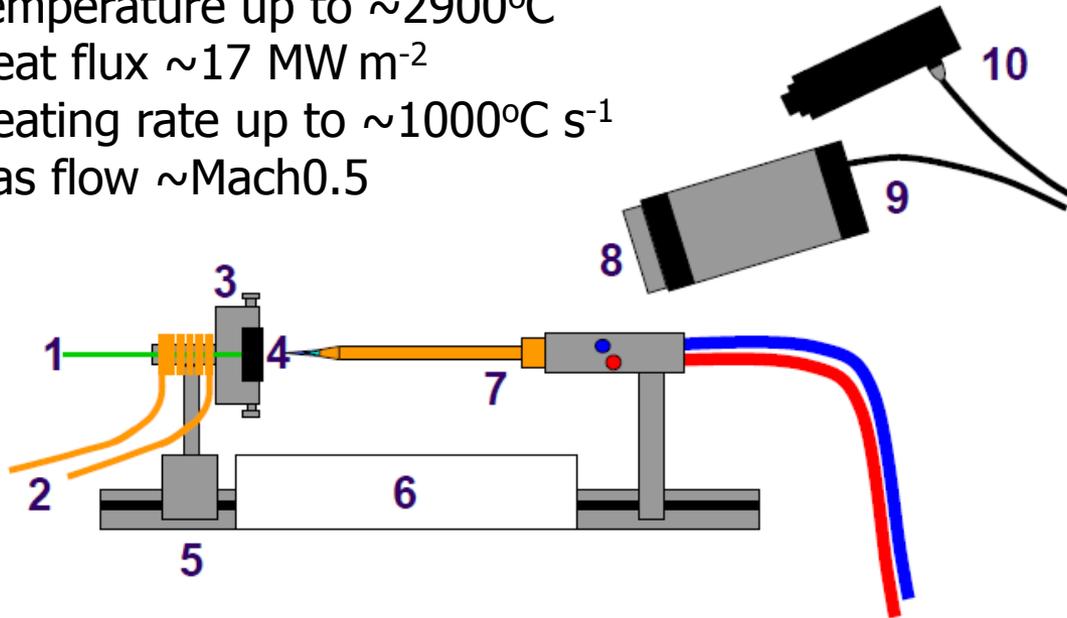


1. Back face thermocouple
2. Water cooling
3. Graphite sample holder
4. Sample
5. Guide rail
6. Protective insulation
7. Oxyacetylene torch
8. Neutral density filter
9. Thermal imaging camera
10. Two colour pyrometer

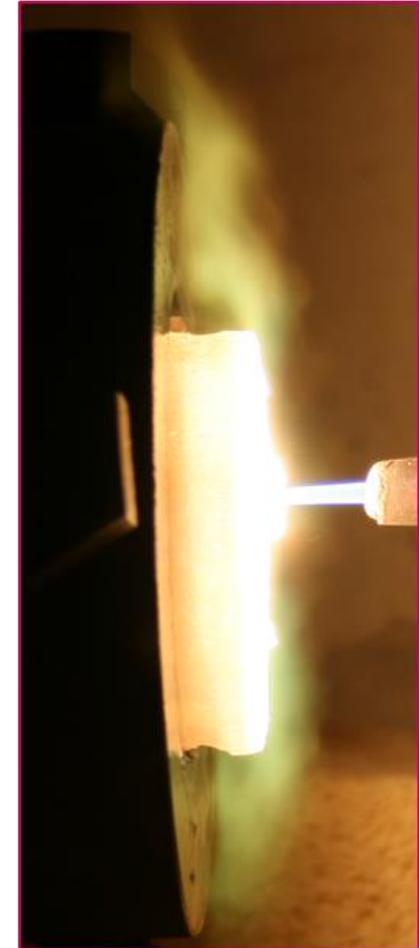
Ability to measure front face temperature,
temperature distribution and back face temperature.

Oxyacetylene Torch Testing (OAT)

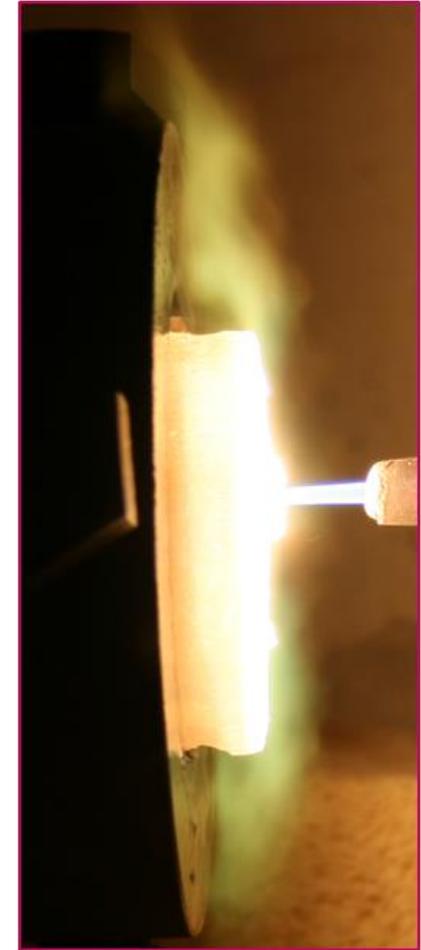
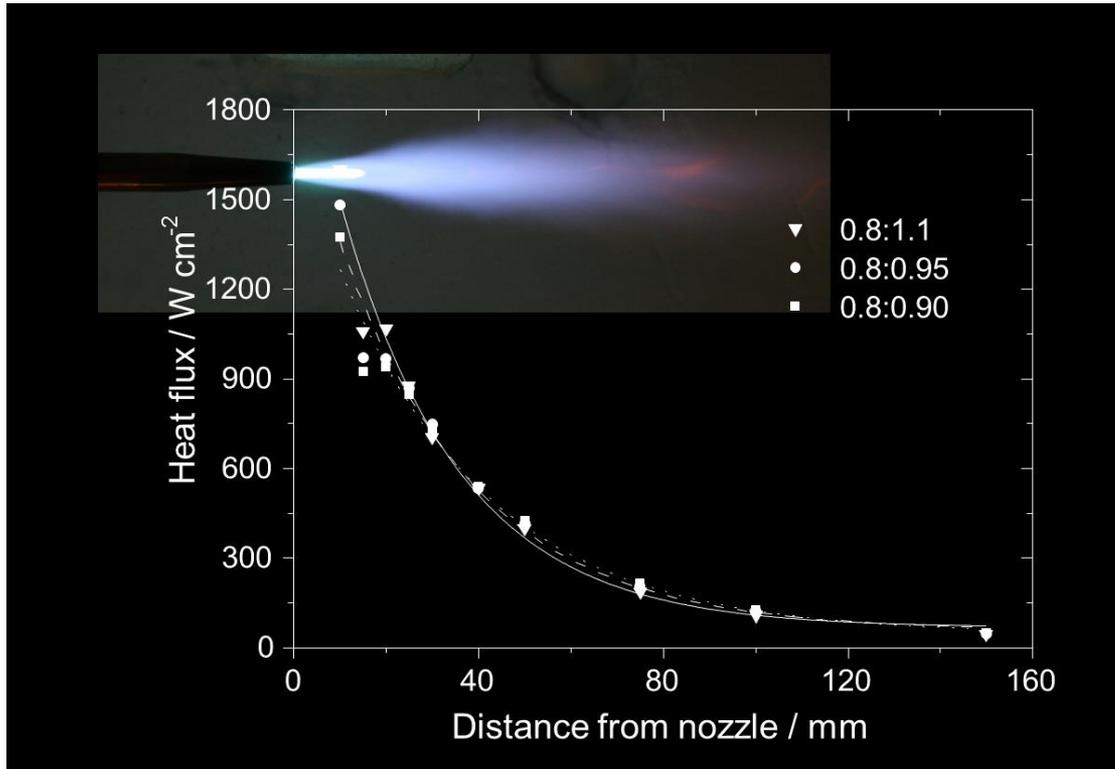
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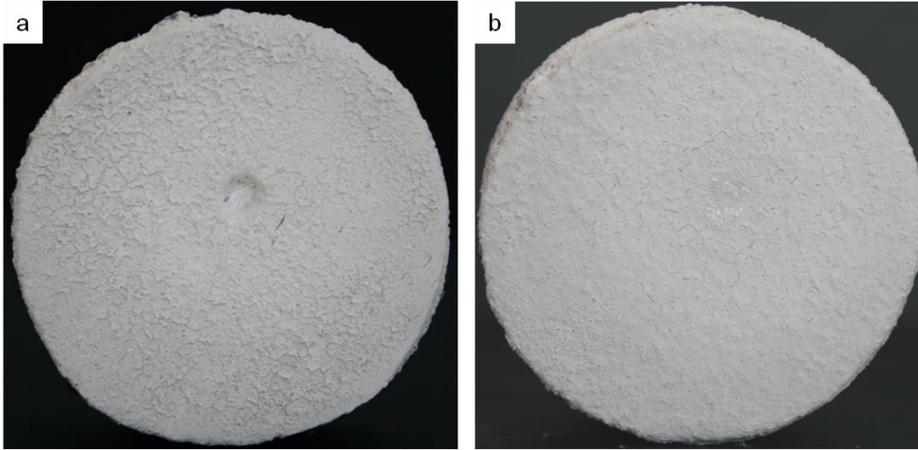
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Oxyacetylene Torch Testing (OAT)

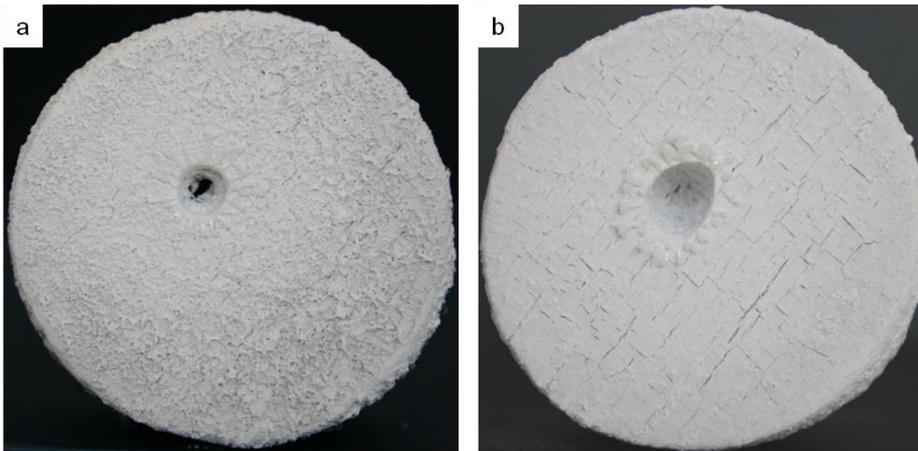


Effect of HfB₂ Particle Size



UHTC composites after 60 s
OA testing @ 2700°C a) Cf-
fine HfB₂ and b) Cf-coarse
HfB₂.

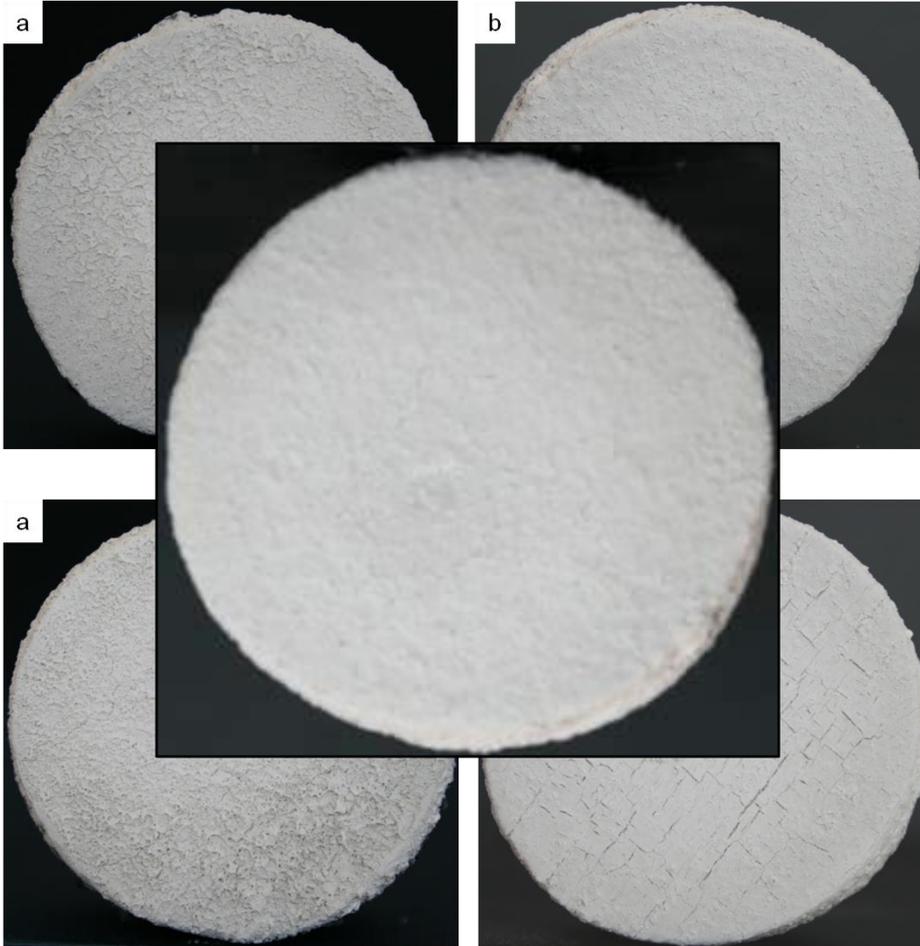
Coarse is better.



UHTC composites after 140 s
OA testing @ 2700°C a) Cf-
fine HfB₂ and b) Cf-coarse
HfB₂.

Fine is better.

Effect of HfB₂ Particle Size



UHTC composites after 60 s
OA testing @ 2700°C a) Cf-
fine HfB₂ and b) Cf-coarse
HfB₂.

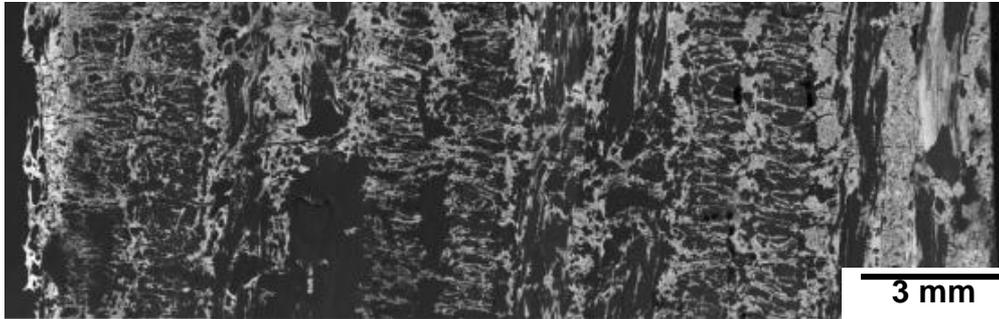
Coarse is better.

The combination is best

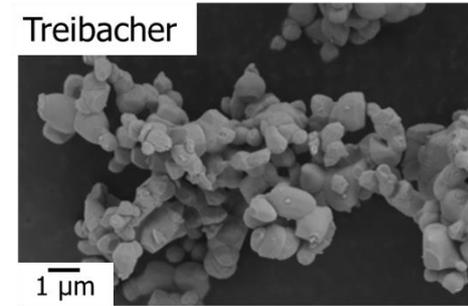
UHTC composites after 140 s
OA testing @ 2700°C a) Cf-
fine HfB₂ and b) Cf-coarse
HfB₂.

Fine is better.

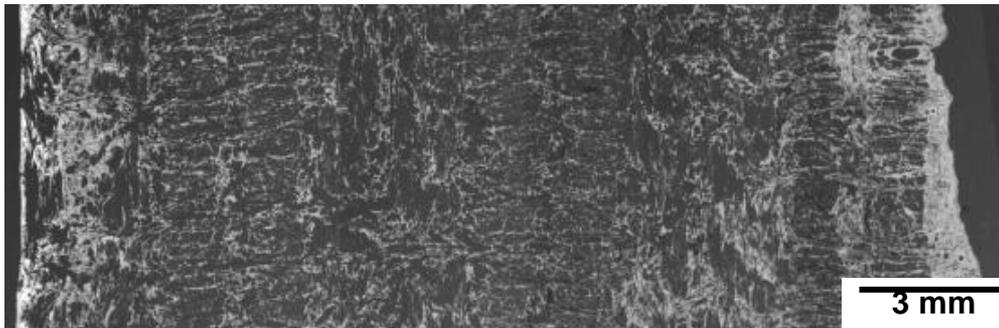
Effect of HfB₂ Particle Size



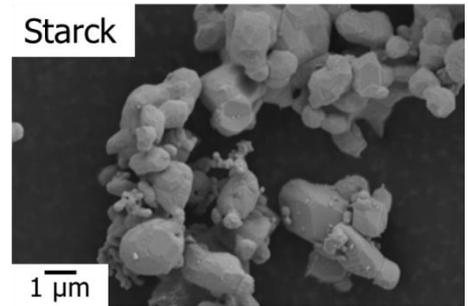
HfB₂ Fine



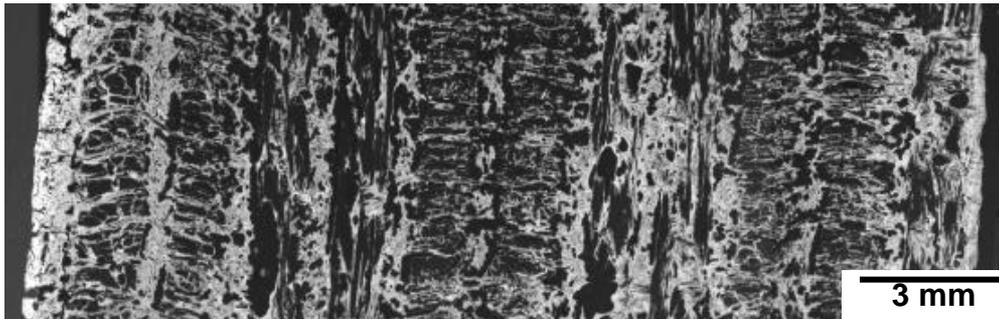
Treibacher



HfB₂ Coarse



Starck

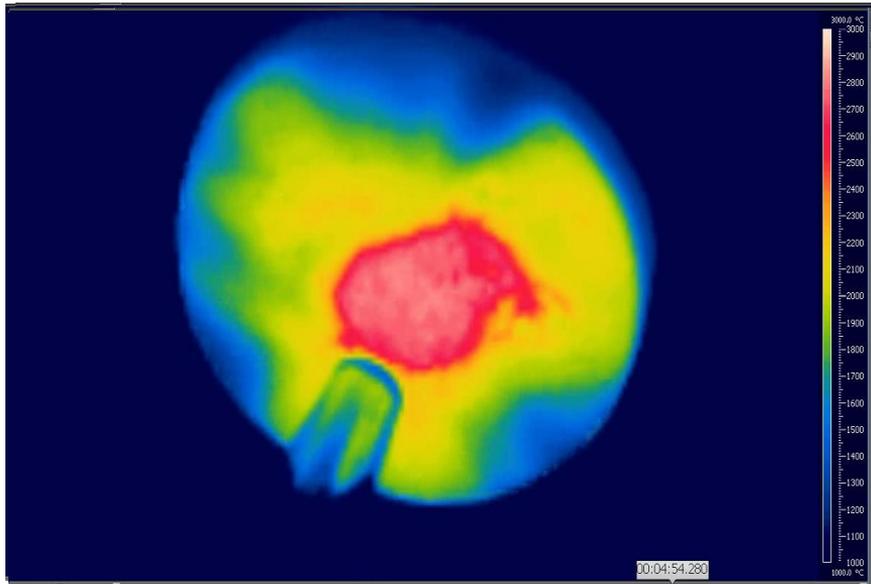


HfB₂ 30% fine & 70% coarse

Oxyacetylene Torch (OAT) Test

Duration: 300 s

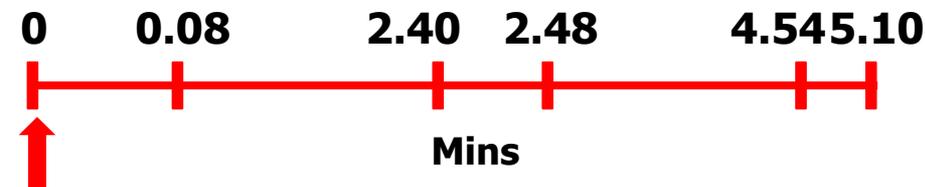
C_2H_2/O_2 0.8 : 1.1 $m^3 h^{-1}$
Heat flux $>17 MW m^{-2}$



At the end of OAT



Sample after
300 s OAT @ $\sim 2900^\circ C$
Mass loss: 2.1 g
Erosion depth:
 ~ 5 mm

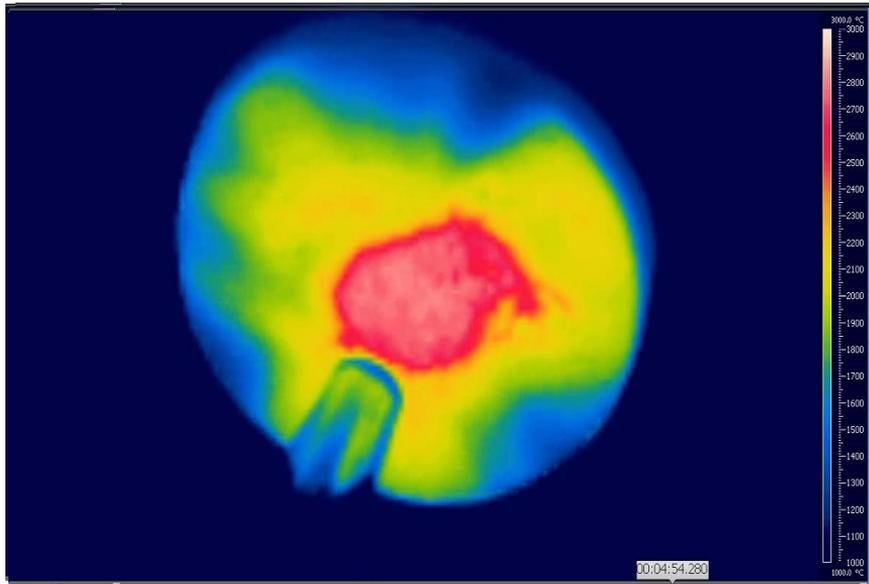


Thermal imaging videos of C_f-HfB_2 samples (edited time frame)

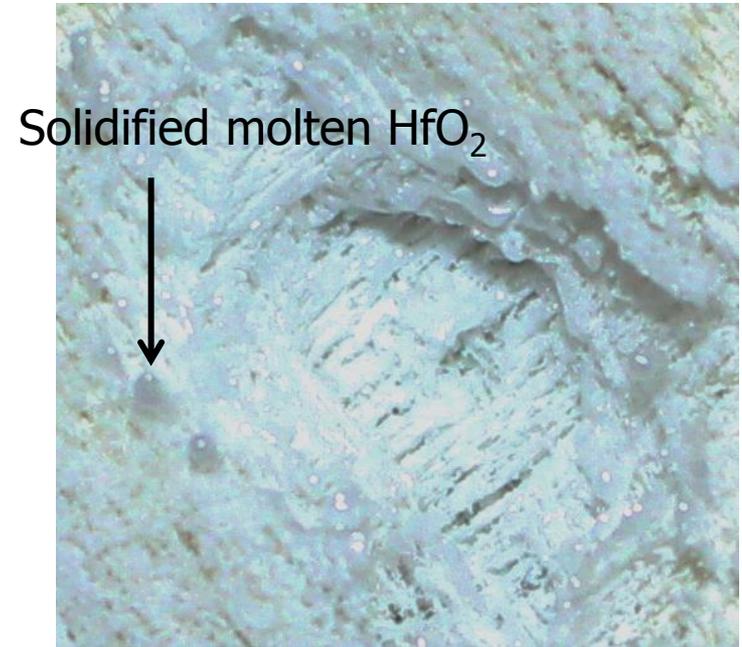
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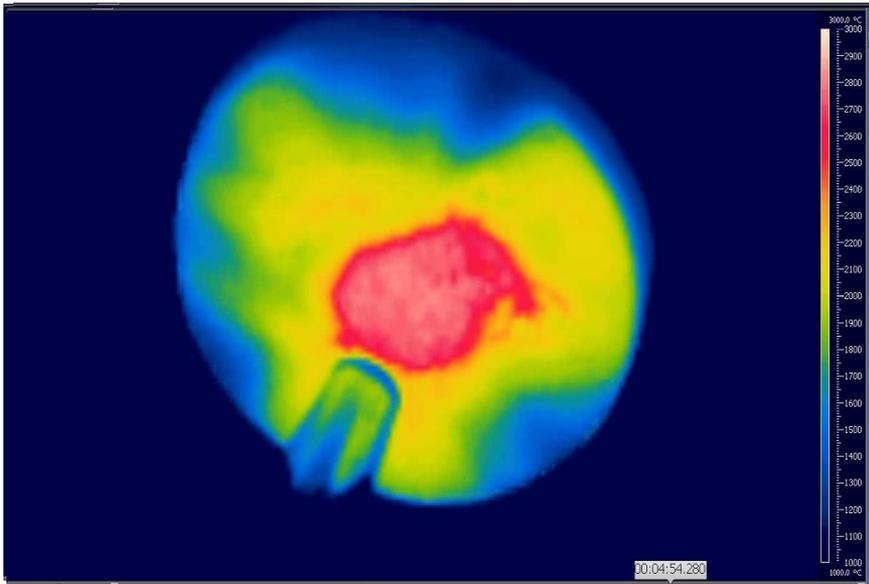


Thermal imaging videos of C_f-HfB_2 samples (edited time frame)

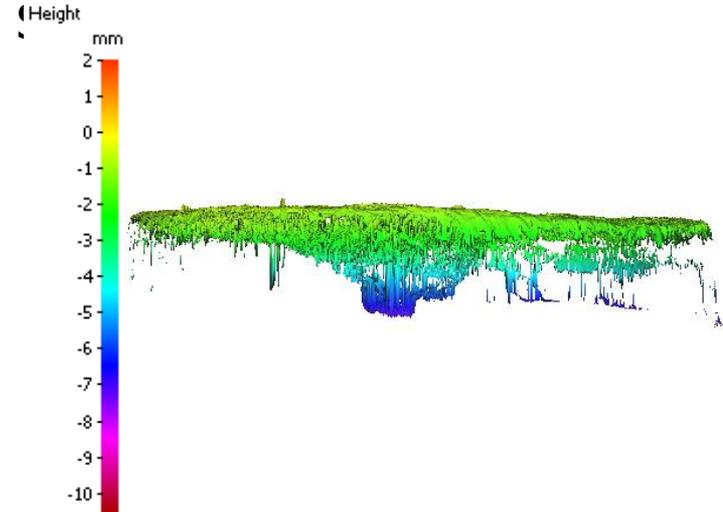
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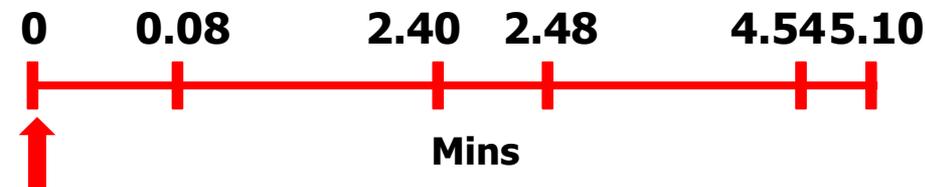
At the end of OAT



300 s OAT @ $\sim 2900^{\circ}C$

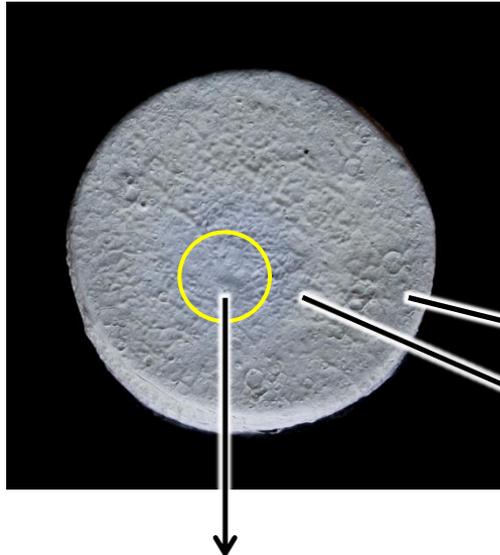
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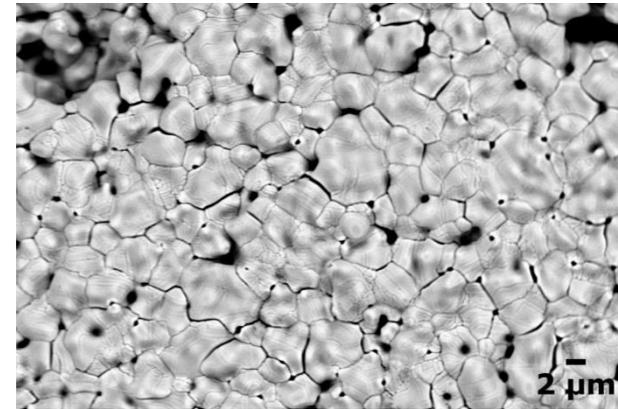
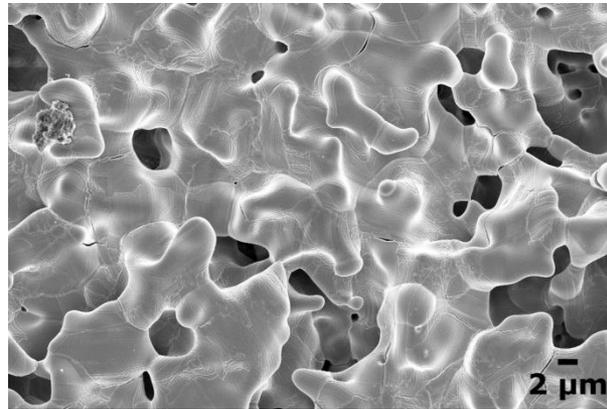
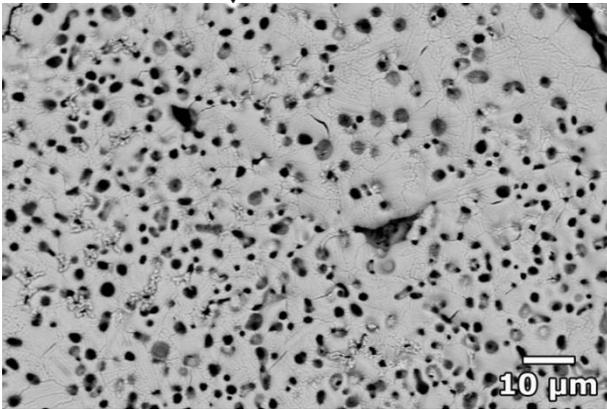


Thermal imaging videos of C_f-HfB_2 samples (edited time frame)

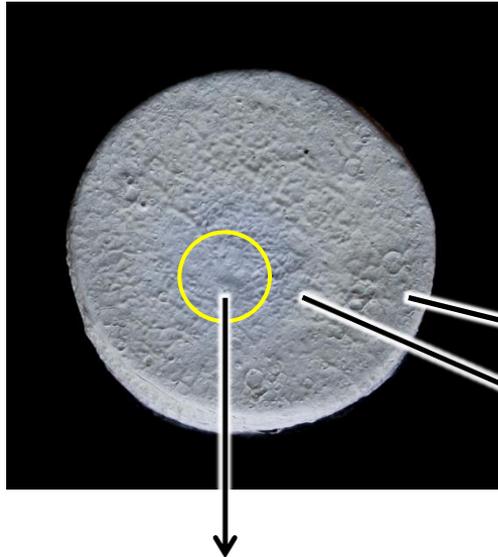
Oxyacetylene Torch Testing (OAT) – 17 MW m⁻² 60 s



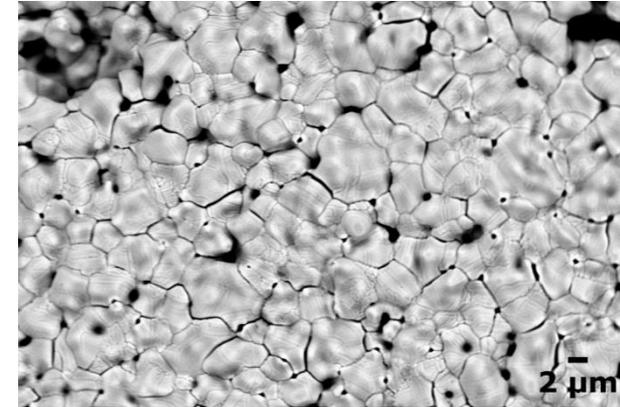
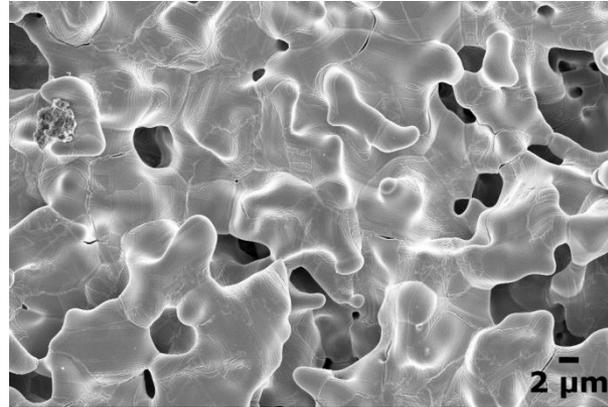
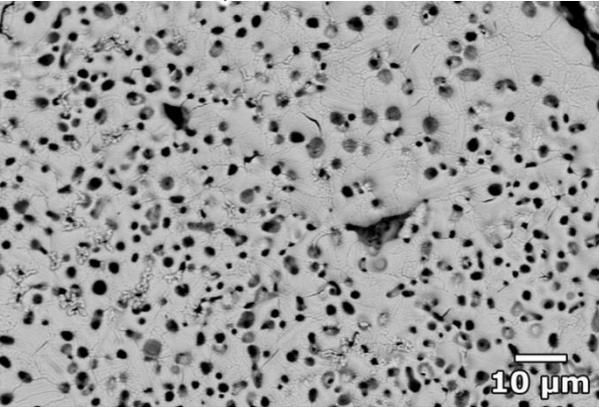
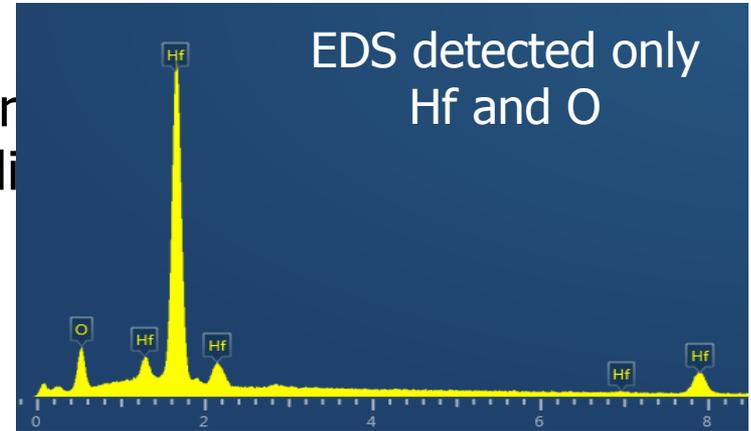
The difference in microstructure is an indication of the temperature gradients across the surface of the sample



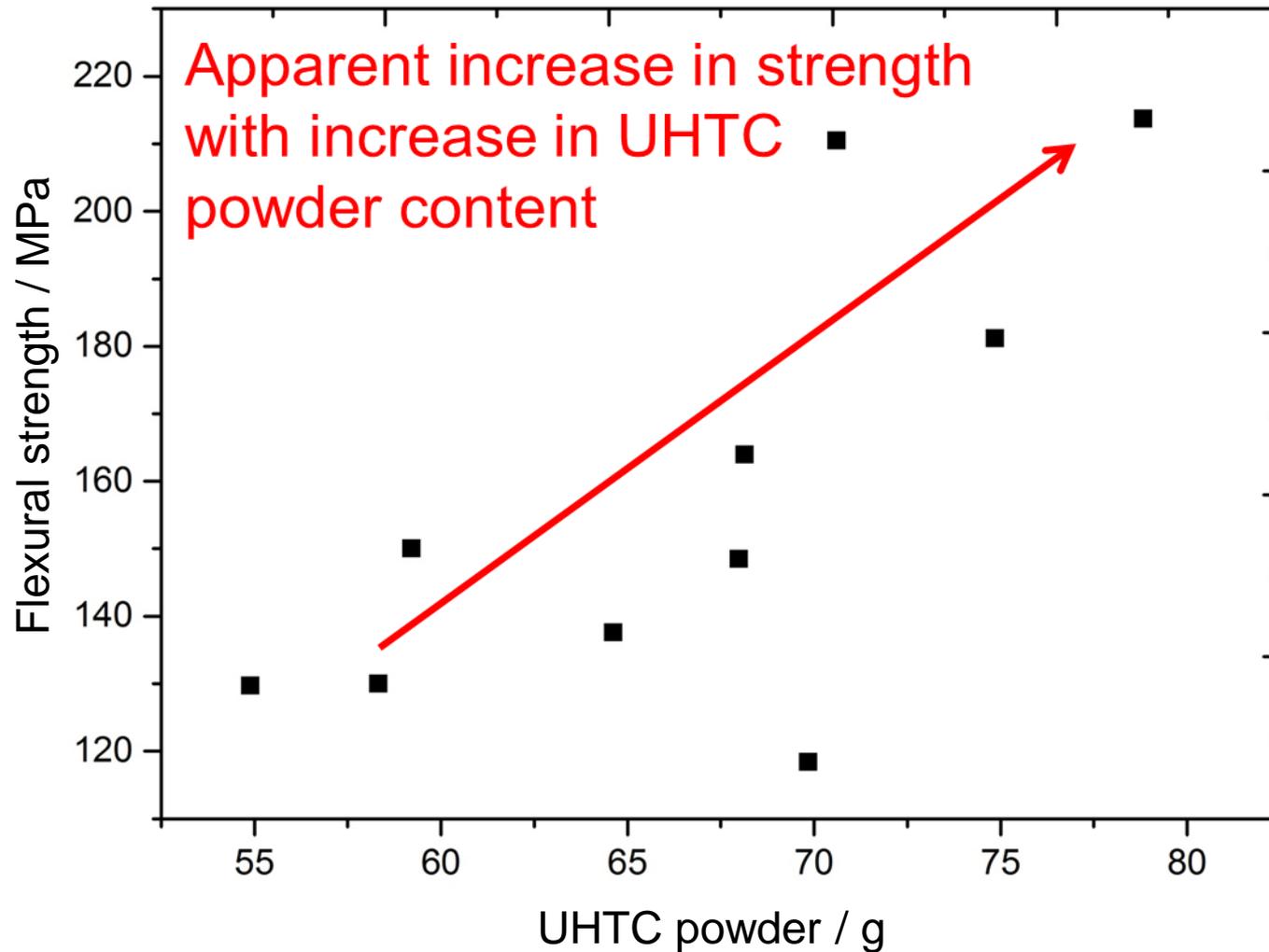
Oxyacetylene Torch Testing (OAT) – 17 MW m⁻² 60 s



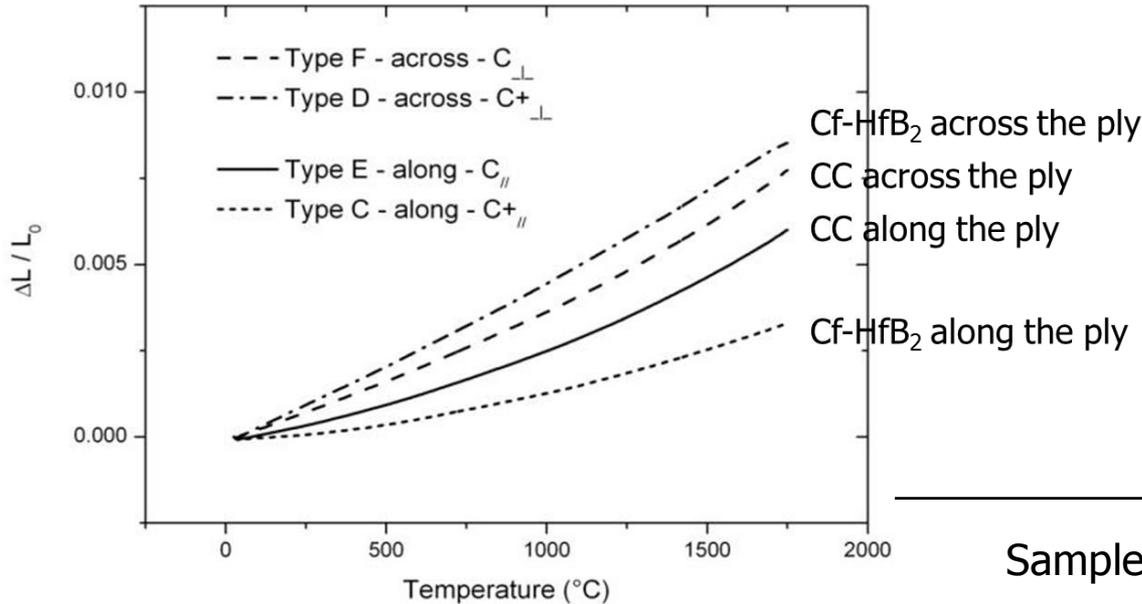
The difference in microstructure due to the temperature gradient in the sample



Flexural Strength vs Final Density



Thermal Expansion



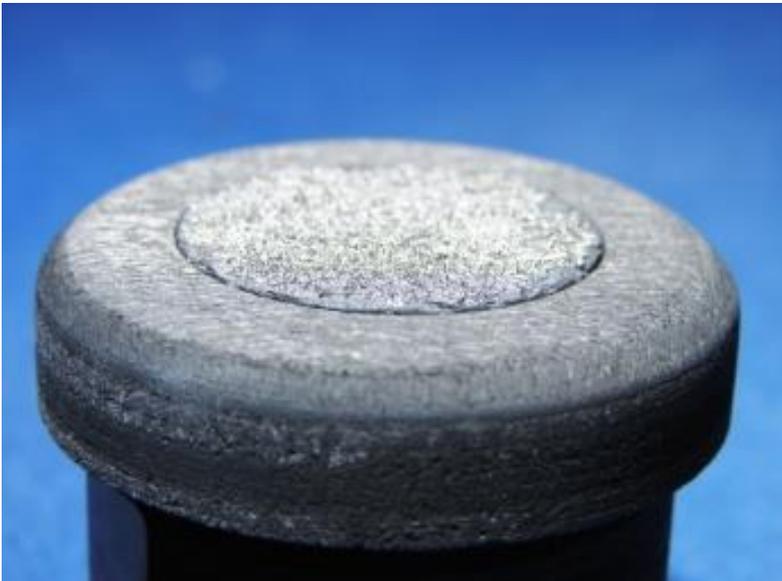
Measured at
Imperial College

The variation along and across the ply will need to be taken into account when designing components.

Sample	Temperature range	Average CTE ($\times 10^{-6} \text{ K}^{-1}$)
CC along the ply	25 – 1700°C	2.23 ± 0.01
CC across the ply	25 – 1700°C	3.92 ± 0.01
Cf-HfB ₂ along the ply	25 – 1700°C	1.77 ± 0.01
Cf-HfB ₂ across the ply	25 – 1700°C	4.70 ± 0.01

Arc Jet Testing

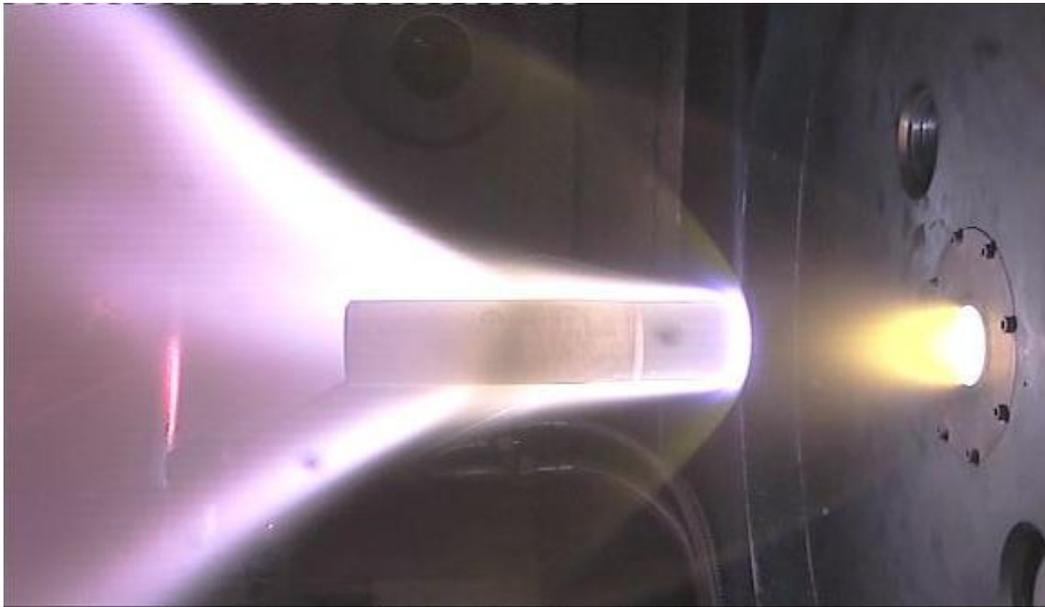
- 5 MW m⁻² for 20 s; Peak T ~2200°C
- 10 MW m⁻² for 10 s; Peak T ~2700°C
- Diameter of the jet ~100 mm
- 30 mm dia x 5 mm thick samples



Before

Arc Jet Testing

- 5 MW m⁻² for 20 s; Peak T ~2200°C
- 10 MW m⁻² for 10 s; Peak T ~2700°C
- Diameter of the jet ~100 mm
- 30 mm dia x 5 mm thick samples



5 MW m⁻² for 20 s



After

10 MW m⁻² for 10 s

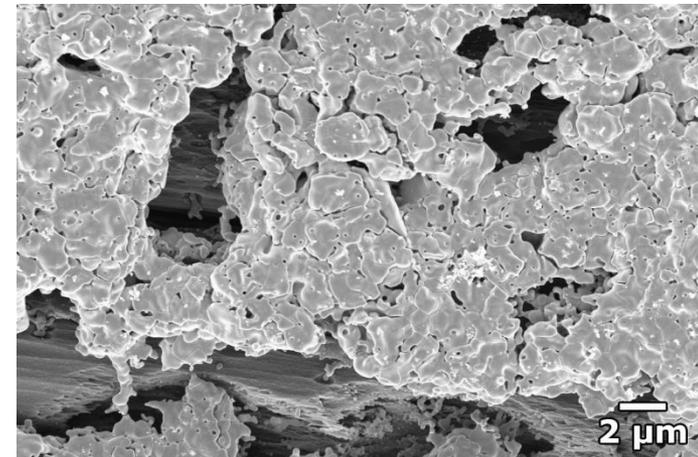
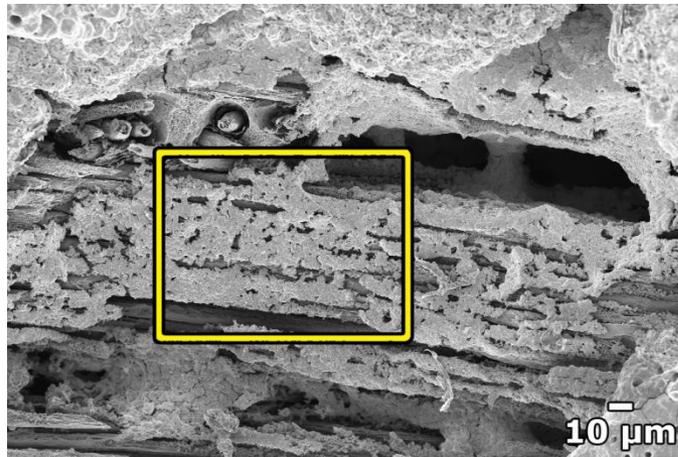
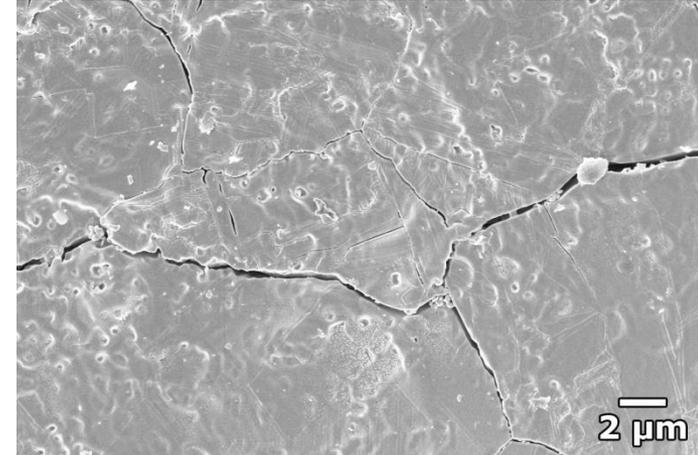
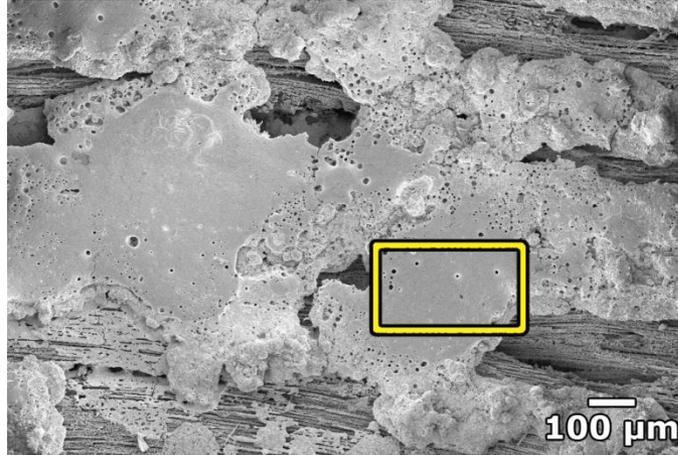


Arc Jet Testing – 10 MW m⁻² 10 s



Notable melting

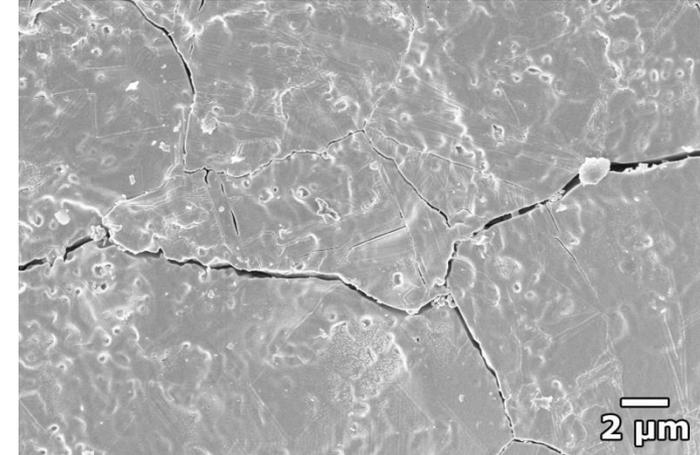
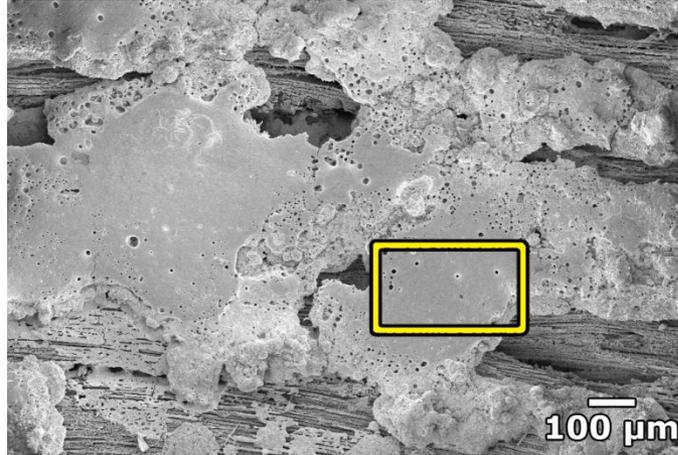
The UHTC phase is protecting the fibres.



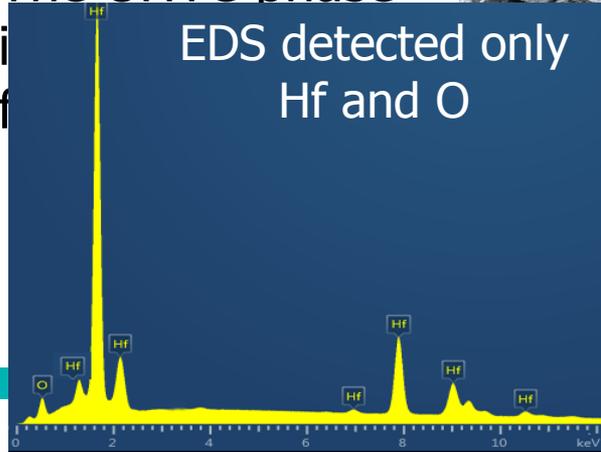
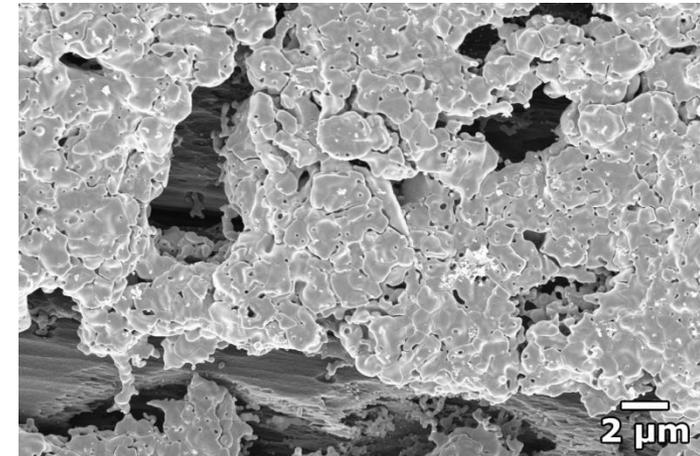
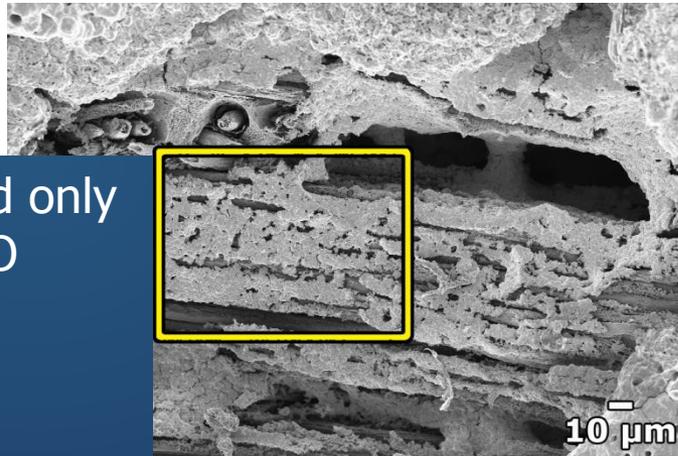
Arc Jet Testing – 10 MW m⁻² 10 s



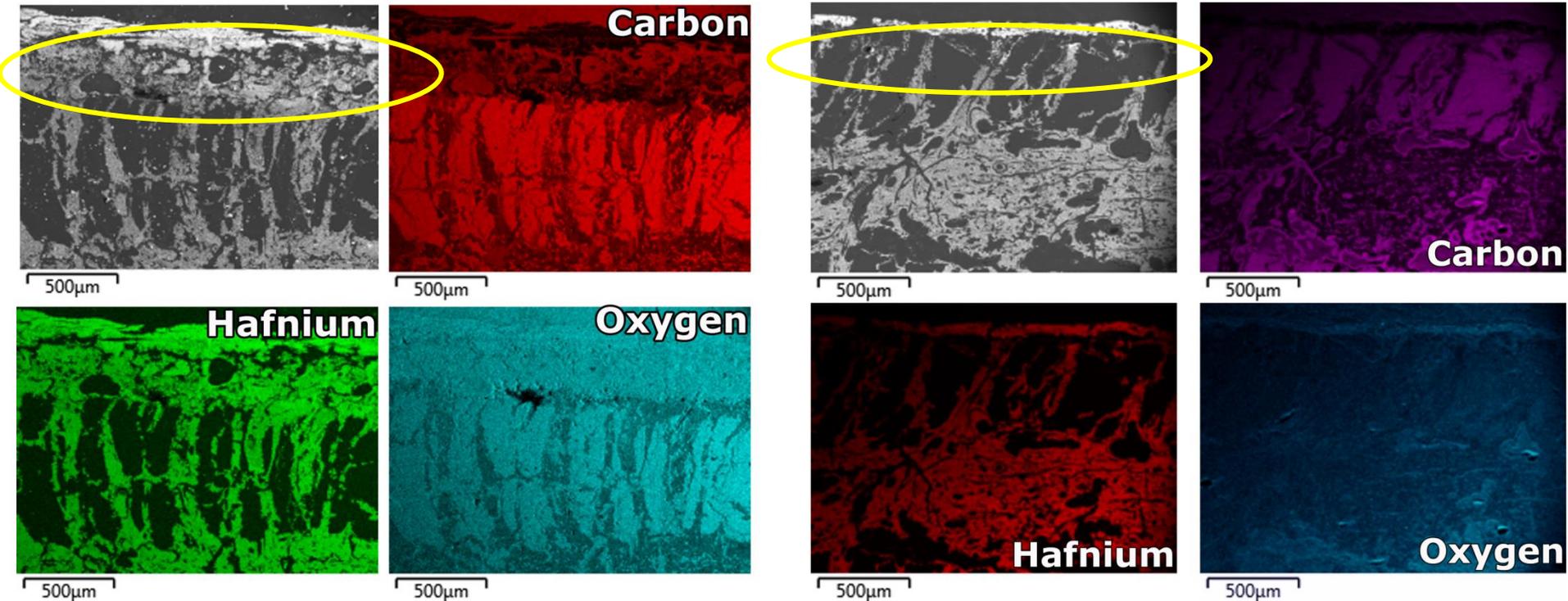
Notable melting



The UHTC phase



Oxyacetylene and Arc Jet Cross Section – EDS

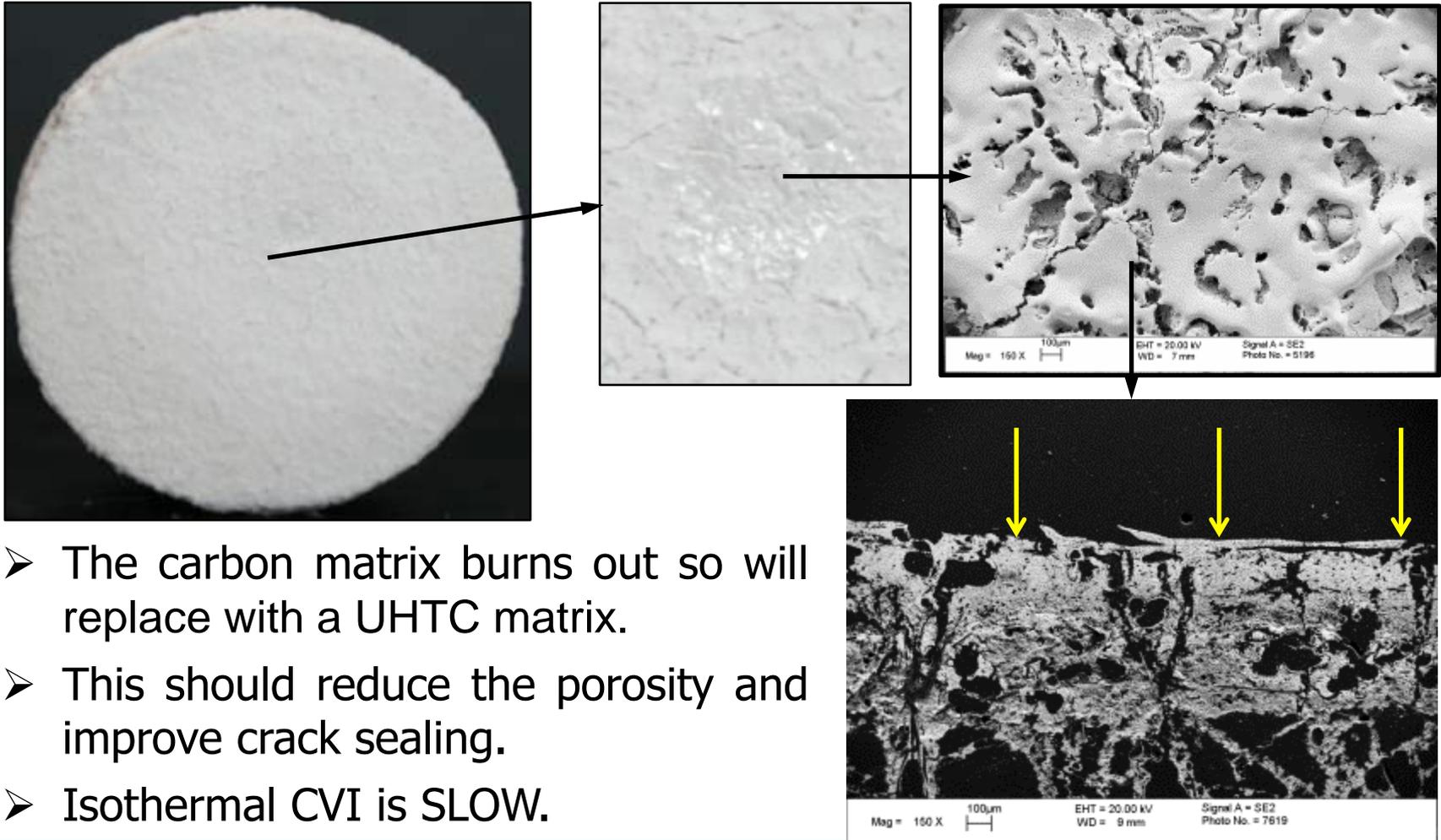


Oxyacetylene 17 MW m⁻² 60 s

Arc-jet 10 MW m⁻² 10 s

Oxidation only observed near the surface layer; depth is ~150 µm for the OAT sample and ~45 µm for the arc-jet sample.

Development Of Porosity During Thermal Testing



- The carbon matrix burns out so will replace with a UHTC matrix.
- This should reduce the porosity and improve crack sealing.
- Isothermal CVI is SLOW.

Chemical Vapour Infiltration (CVI)

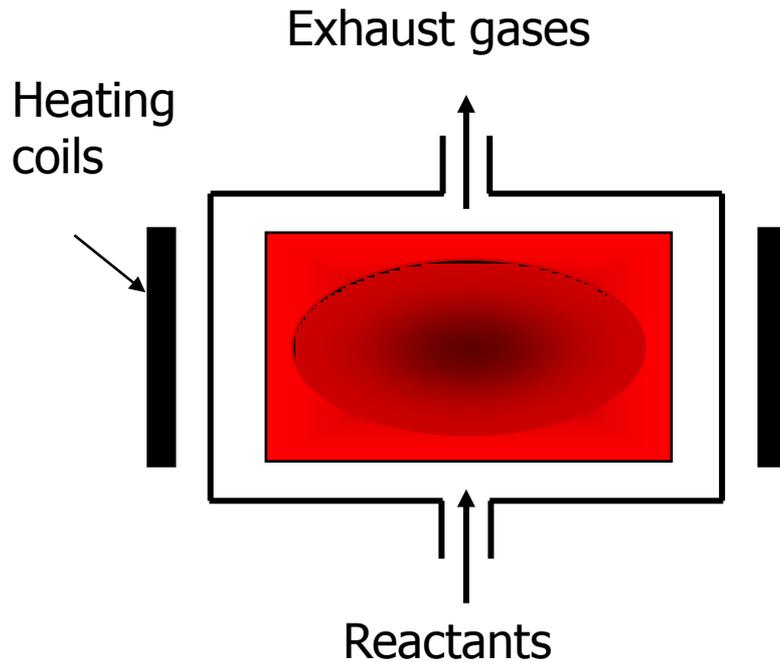
Advantages

- Near net shape
- Wide range of compositions
- Processing T typically $<1000^{\circ}\text{C}$
- Matrix is pure and fine grained
- Near-zero porosity possible
- Can deposit interfacial layers *in-situ* to enhance fibre pullout

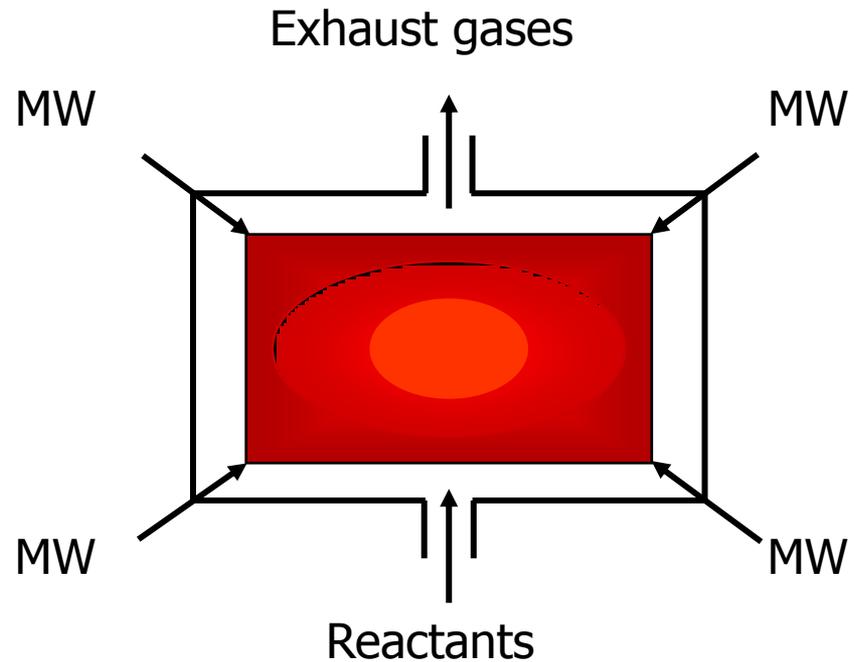
Disadvantage

- Conventional isothermal, isobaric (I-CVI) process, very slow
- ⇒ Many variations:
 - pressure & concⁿ gradient
 - temperature gradient
 - forced flow (F-CVI)
 - pulsed-flow (P-CVI)
 - *microwave heated (M-CVI)*

Conventional vs Microwave CVI

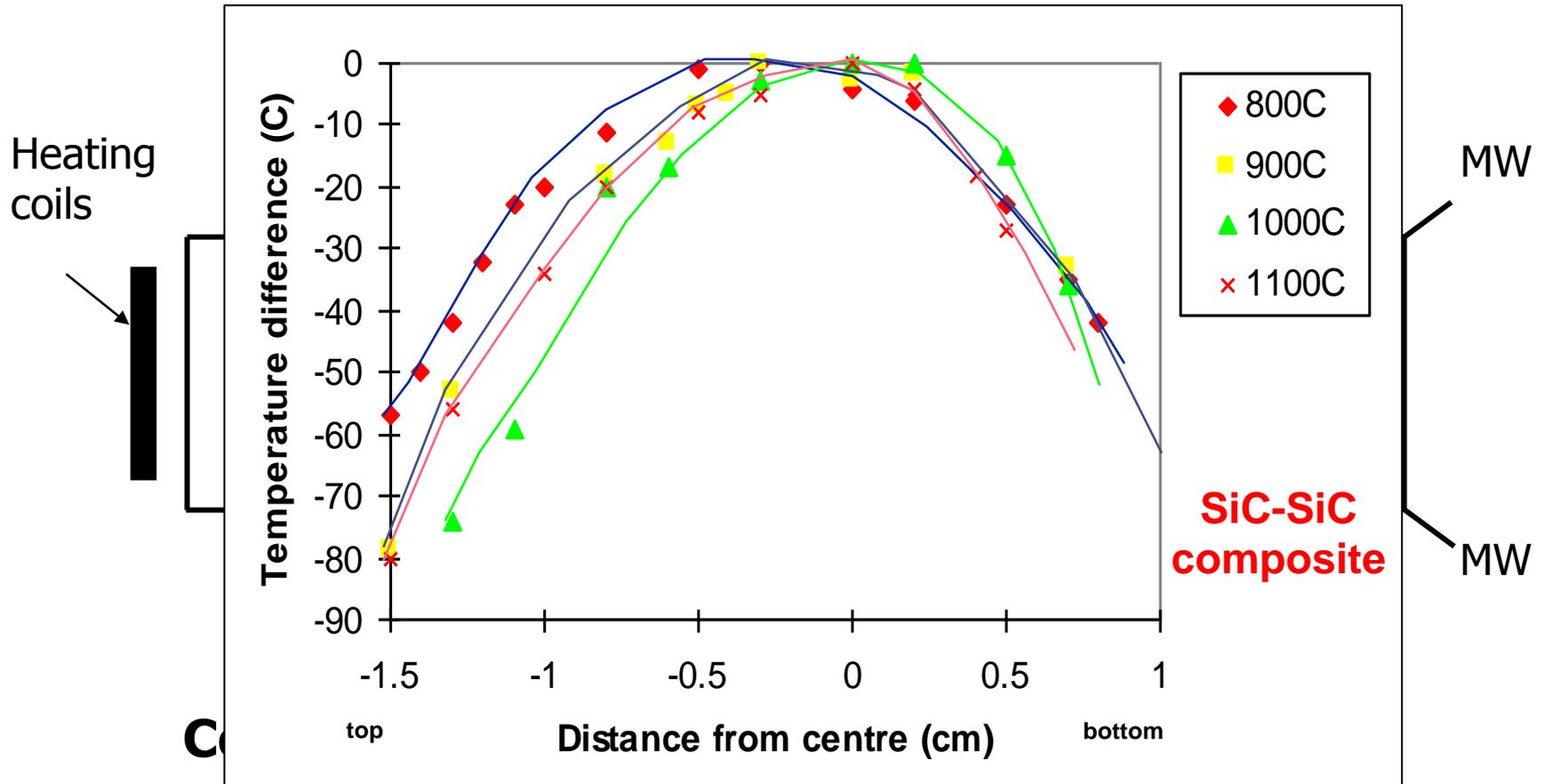


Conventional CVI

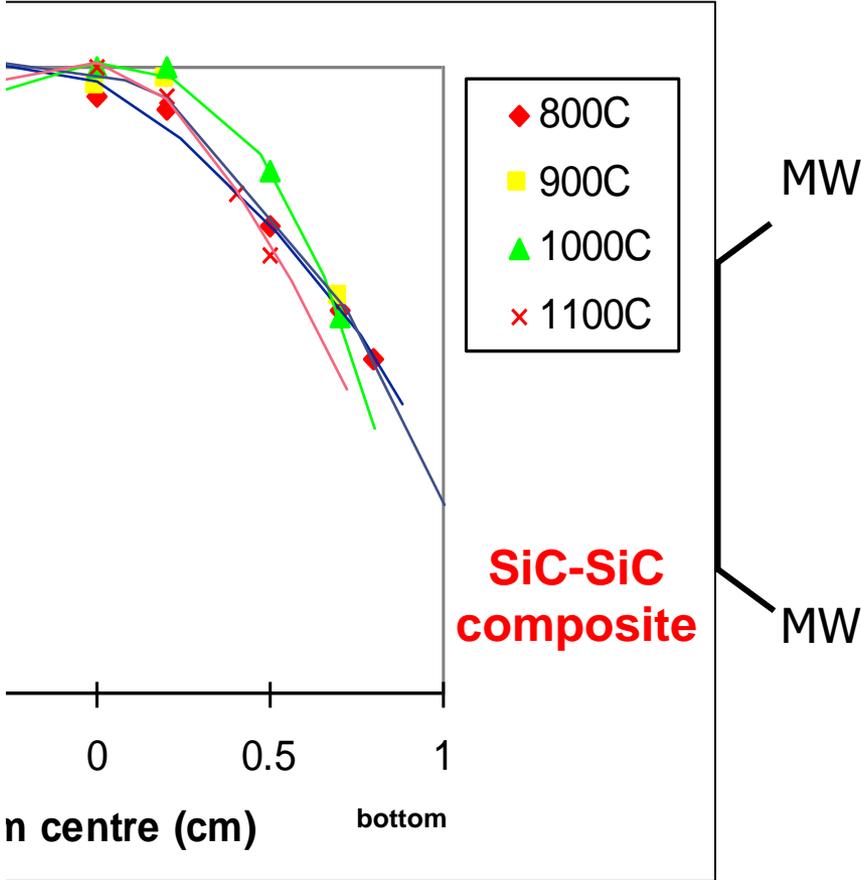


Microwave CVI

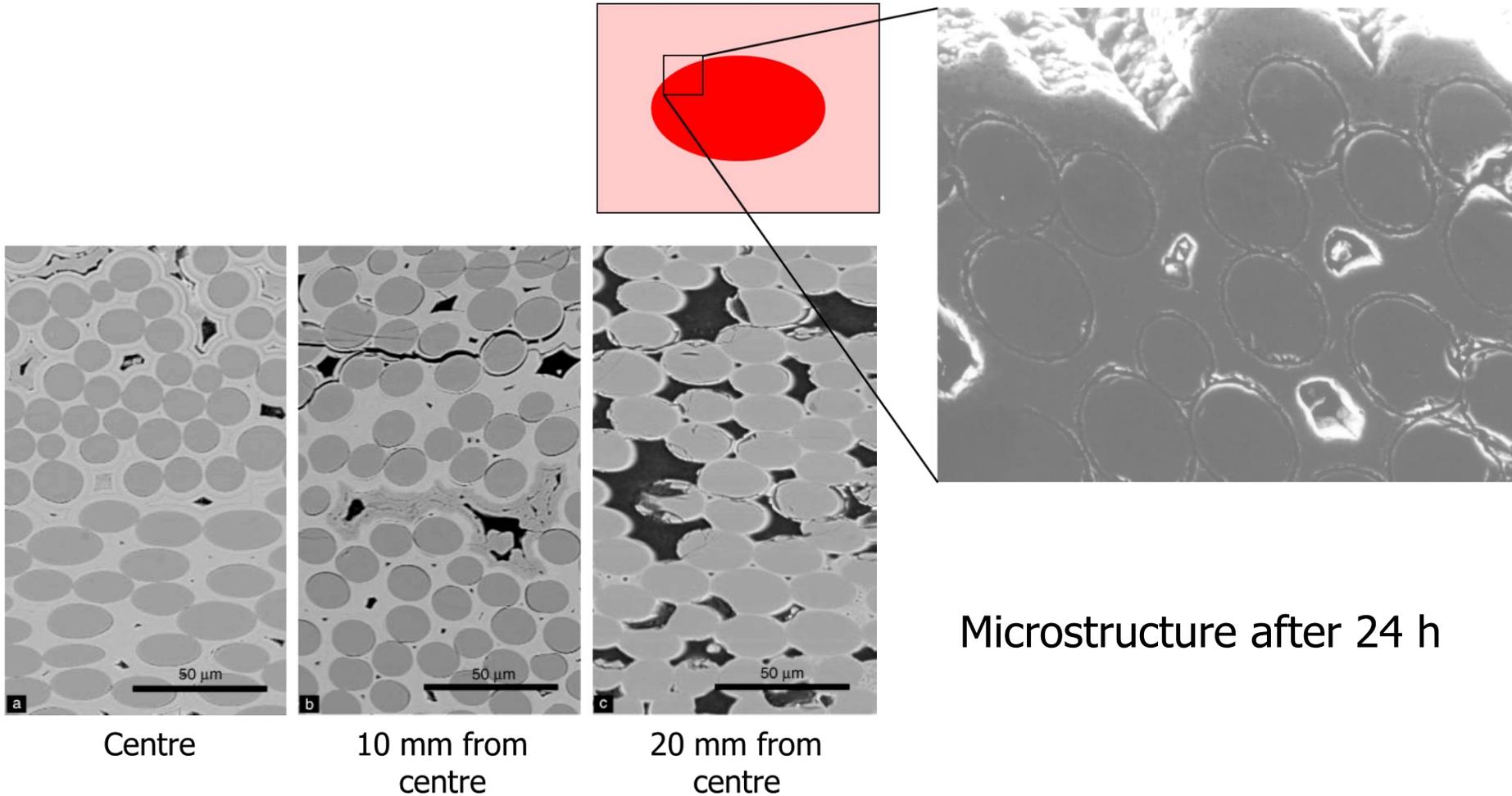
Conventional vs Microwave CVI



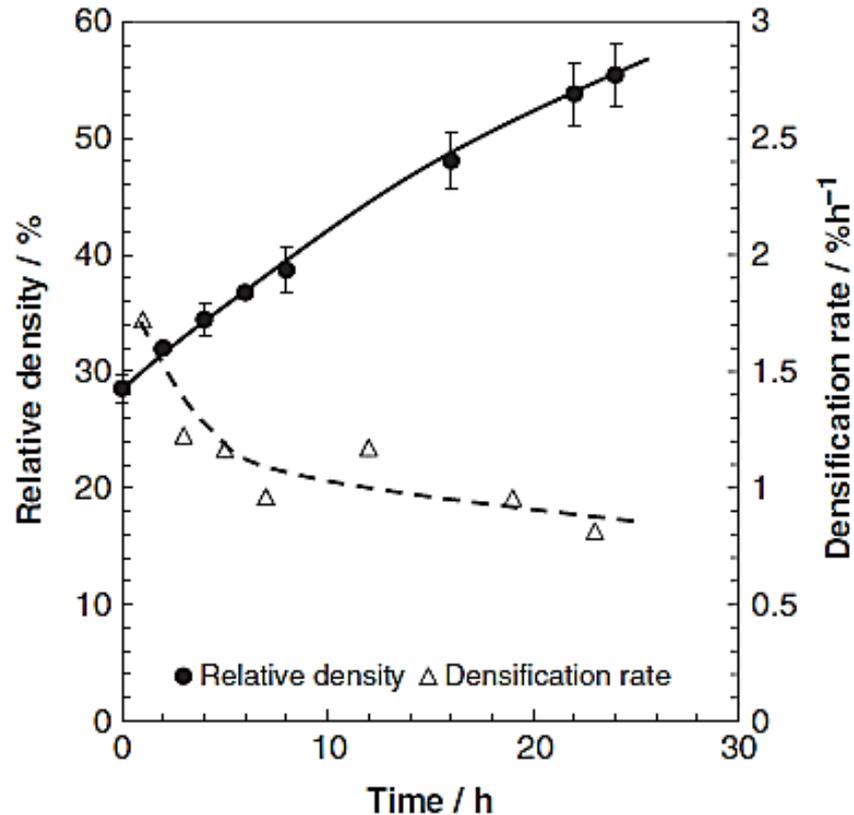
Conventional vs Microwave CVI



Microwave CVI – SiC-SiC Composites



Microwave CVI – Densification Rate



Density vs Time

- Densification occurs from the inside out, hence no crusting.
- Almost fully dense matrix with no density gradients.
- Cold zone restricted to surface, which can be machined off or infiltrated conventionally.
- Opportunity to produce fibre reinforced CMCs in 48 – 72 h instead of 3 months.
- Plan is to repeat for C_f-HfB₂ composites with UHTC matrices.

Summary and Where Next

- C_f-HfB₂ composites show excellent thermal shock, ultra-high temperature and ablation resistance when either oxyacetylene torch or arc jet tested.
- Variability in strength has been addressed by the development of a new impregnation technique.
- Future work is focused on reducing the porosity at high temperature, making both larger (325 x 325 x 5 mm) and more complex-shaped components.
- Final components are probably going to consist of joined combinations of composites and monolithics; work has recently started on this topic too.

Thank You

Acknowledgements

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