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Effect of steam on the performance of Ca-based sorbents in calcium looping processes

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Antonio Coppola, Eduardo Gais, Gabriella Mancino, Fabio Montagnaro, Fabrizio Scala, Piero Salatino

Overview: the Ca-looping concept



Overview: the Ca-looping concept

Sorbent-related Issues (1/2)







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Overview: the Ca-looping concept

Sorbent-related Issues (2/2)



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Overview: effect of steam



Overview: effect of steam



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Experimental

Lab-scale Fluidized bed (40mm-ID) CO_2 Capture Capacity (ξ)German Limestone (EnBW) (0.4-0.6 μ m)Fragmentation tendency4 complete cycles + 5° Calcination N_2 porosimetry + SEM

	calcination stages	carbonation stages
dry		
Temperature	940°C	650°C
Test duration	20 min	15 min
Fluidization superficial velocity	0.7 m s ⁻¹	0.6 m s ⁻¹
Fluidizing gas composition (vol.)	$70\% CO_2 + 30\%$ air	15% CO ₂ +85% air
ste_cal		
Fluidizing gas composition (vol.)	10% steam+70% CO ₂ +20% air	15% CO ₂ +85% air
ste_car		
Fluidizing gas composition (vol.)	70% CO ₂ +30% air	10% steam+15% CO ₂ +75% air
ste_cal_car		
Fluidizing gas composition (vol.)	10% steam+70% CO ₂ +20% air	10% steam+15% CO ₂ +75% air
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Conclusions

- 🝁 steam is beneficial (CO₂ uptake)
- ϕ calcination development of accessible porosity CO₂ uptake in the order of 10%
- ***** carbonation positive role of steam as a "catalyst" of CO₂ diffusion through the sorbent CaCO₃-based product layer.
- 🝁 calcination + carbonation: synergistic effects
- fragmentation propensity: during calcination induces a more resistant external particle structure.

results highlight the positive role that the presence of steam in realistic calcium looping conditions

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Thank you for your attention

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